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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.054$
$w R$ factor $=0.125$
Data-to-parameter ratio $=15.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## $N$-(4-Methoxybenzoyl)- $\mathrm{N}^{\prime}$-[2-(morpholinium-1-yl)ethyl]thiourea thiocyanate

The title compound \{systematic name: 4-[2-(\{[(4-methoxy-benzoyl)amino]carbonothioyl\}amino)ethyl]morpholin-4-ium thiocyanate\}, $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+} \cdot \mathrm{NCS}^{-}$, is a $p$-anisoylthiourea salt with a protonated morpholine group and a thiocyanate counter-anion. The morpholinium group has a chair conformation and the dihedral angle between the $p$-anisoyl group and the central thiourea $\mathrm{N}_{2} \mathrm{CS}$ group is 32.51 (8) ${ }^{\circ}$. The crystal structure is stablized by intra- and intermolecular hydrogen bonds.

## Comment

Most anisoylthiourea derivatives have a trans-cis configuration with respect to the thiono group across their $\mathrm{C}-\mathrm{N}$ bonds such as in $N$-(1-benzylpiperidine-4-yl)- $N^{\prime}$-(4-methoxybenzoyl)thiourea (Yusof \& Yamin, 2005). The title compound, (I), is also an anisoylthiourea derivative that has similar properties, but the morpholinium-1-ylethyl group bound to the terminal atom N 3 is protonated. The charged thiourea molecule is balanced by the thiocyanate counter-anion (Fig.1).

(I)

The bond lengths and angles are in normal ranges (Allen et al., 1987) and in agreement with those in $N$-(1-benzyl-piperidine-4-yl)- $N^{\prime}$-(4-methoxybenzoyl)thiourea. The thiocyanate $\mathrm{C} 16-\mathrm{S} 2$ bond length is slightly shorter than the thiono $\mathrm{C} 8=\mathrm{S} 1$ (Table 1). The $p$-anisoyl fragment [C7/C1-C6/ O2/C15, maximum deviation 0.019 (2) $\AA$ at C6] is inclined to the central thiourea group $[\mathrm{N} 1 / \mathrm{C} 8 / \mathrm{S} 1 / \mathrm{N} 3 / \mathrm{C} 9$, maximum deviation $0.022(2)^{\circ}$ at C9] by $32.51(8)^{\circ}$. There are two intramolecular hydrogen bonds, $\mathrm{N} 2-\mathrm{H} 2 A \cdots \mathrm{O} 1$ and $\mathrm{C} 9-$ $\mathrm{H} 9 A \cdots \mathrm{~S} 1$, resulting in the formation of a six-membered ring $(\mathrm{H} 2 A \cdots \mathrm{O} 1-\mathrm{C} 7-\mathrm{N} 1-\mathrm{C} 8-\mathrm{N} 2)$ and a five-membered ring (H9A $\cdots \mathrm{S} 1-\mathrm{C} 8-\mathrm{N} 2-\mathrm{C} 9)$, respectively. The $\mathrm{N} 3-\mathrm{H} 3 A \cdots \mathrm{~N} 4$ hydrogen bond links the cation to the counter-anion. In the crystal structure, there are intermolecular hydrogen bonds of the type $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}, \mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{S}$ (Table 2) to form a polymeric chain network (Fig. 2).

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## Experimental

A solution of 2-(morpholine-4-yl)ethylamine ( $0.53 \mathrm{~g}, 4.1 \mathrm{mmol}$ ) in acetone $(50 \mathrm{ml})$ was added dropwise to an acetone solution $(50 \mathrm{ml})$ containing an equimolar amount of $p$-methoxybenzoyl isothiocyanate in a 200 ml two-necked round-bottomed flask. The solution was refluxed for 2 h and then poured into a beaker containing some crushed ice. The white precipitate was washed with cold ethanol and distilled water, then dried in a vacumm (yield 80\%). Recrystallization from acetone yielded single cystals suitable for X-ray analysis (m.p. 417.1-419.0 K). Analysis calculated: C 50.24, H 5.80, N 14.65, S 16.77, O 12.55\%; found: C 49.3, H 5.60, N 14.13, S 15.86, O 11.983\%.

## Crystal data

$\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}^{+} \cdot \mathrm{NCS}^{-}$
$M_{r}=382.50$
Monoclinic, $P 2_{1} / n$
$a=9.0195$ (18) £
$b=10.402$ (2) A
$c=20.289$ (4) $\AA$
$\beta=97.477$ (3) ${ }^{\circ}$
$V=1887.3(7) \AA^{3}$
$Z=4$
$D_{x}=1.346 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 4016 reflections
$\theta=2.0-25.5^{\circ}$
$\mu=0.31 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, colourless
$0.42 \times 0.33 \times 0.25 \mathrm{~mm}$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.882, T_{\text {max }}=0.927$
9351 measured reflections
3505 independent reflections
3107 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.018$
$\theta_{\text {max }}=25.5^{\circ}$
$h=-10 \rightarrow 10$
$k=-5 \rightarrow 12$
$l=-23 \rightarrow 24$

$$
\begin{aligned}
& w=1 / {\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0429 P)^{2}\right.} \\
&+0.9468 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }<0.00 \\
& \Delta \rho_{\max }=0.33 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.27 \mathrm{e}^{-3}
\end{aligned}
$$



Figure 1
A view of the molecule of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level.


Figure 2
The crystal packing of the cation and anion of (I), viewed along the $b$ axis. Dashed lines indicate hydrogen bonds.

After their location in a difference map, all H atoms except H3A were positioned geometrically at ideal positions and allowed to ride on the parent atom, with $\mathrm{C}-\mathrm{H}=0.93-0.96 \AA$ and $\mathrm{N}-\mathrm{H}=0.86 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}\left(\mathrm{CH}_{2}\right.$ and NH$)$ or $1.5 U_{\mathrm{eq}}\left(\mathrm{CH}_{3}\right)$. The protonated atom $\mathrm{H} 3 A$, bonded to N 3 , was located in a difference map and refined freely.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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