Received 29 June 2005 Accepted 4 July 2005

Online 9 July 2005

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

M. Sukeri M. Yusof^a and Bohari M. Yamin^b*

^aDepartment of Chemistry, Kolej Universiti Sains dan Teknologi Malaysia, Mengabang Telipot, 21030 Kuala Terengganu, Malaysia, and ^bSchool of Chemical Sciences and Food Technology, Universiti Kebangsaan Malaysia, 43600 Bangi, Selangor, Malaysia

Correspondence e-mail: bohari@pkrisc.cc.ukm.my

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.054 wR factor = 0.125 Data-to-parameter ratio = 15.2

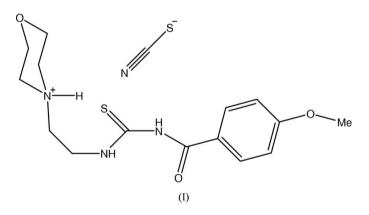
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N-(4-Methoxybenzoyl)-*N*'-[2-(morpholinium-1-yl)ethyl]thiourea thiocyanate

The title compound {systematic name: 4-[2-({[(4-methoxybenzoyl)amino]carbonothioyl}amino)ethyl]morpholin-4-ium thiocyanate}, $C_{15}H_{22}N_3O_3S^+ \cdot NCS^-$, is a *p*-anisoylthiourea salt with a protonated morpholine group and a thiocyanate counter-anion. The morpholinium group has a chair conformation and the dihedral angle between the *p*-anisoyl group and the central thiourea N₂CS group is 32.51 (8)°. The crystal structure is stablized by intra- and intermolecular hydrogen bonds.

Comment

Most anisoylthiourea derivatives have a *trans-cis* configuration with respect to the thiono group across their C—N bonds such as in N-(1-benzylpiperidine-4-yl)-N'-(4-methoxybenzoyl)thiourea (Yusof & Yamin, 2005). The title compound, (I), is also an anisoylthiourea derivative that has similar properties, but the morpholinium-1-ylethyl group bound to the terminal atom N3 is protonated. The charged thiourea molecule is balanced by the thiocyanate counter-anion (Fig.1).



The bond lengths and angles are in normal ranges (Allen et al., 1987) and in agreement with those in N-(1-benzylpiperidine-4-yl)-N'-(4-methoxybenzoyl)thiourea. The thiocvanate C16-S2 bond length is slightly shorter than the thiono C8=S1 (Table 1). The *p*-anisoyl fragment [C7/C1-C6/ O2/C15, maximum deviation 0.019 (2) Å at C6] is inclined to the central thiourea group [N1/C8/S1/N3/C9, maximum deviation 0.022 (2)° at C9] by 32.51 (8)°. There are two intramolecular hydrogen bonds, N2-H2A...O1 and C9-H9A···S1, resulting in the formation of a six-membered ring $(H2A \cdots O1 - C7 - N1 - C8 - N2)$ and a five-membered ring $(H9A \cdots S1 - C8 - N2 - C9)$, respectively. The N3 - H3A \cdots N4 hydrogen bond links the cation to the counter-anion. In the crystal structure, there are intermolecular hydrogen bonds of the type C-H···O, N-H···S and C-H···S (Table 2) to form a polymeric chain network (Fig. 2).

Printed in Great Britain - all rights reserved

© 2005 International Union of Crystallography

Experimental

A solution of 2-(morpholine-4-yl)ethylamine (0.53 g, 4.1 mmol) in acetone (50 ml) was added dropwise to an acetone solution (50 ml) containing an equimolar amount of *p*-methoxybenzoyl isothiocyanate in a 200 ml two-necked round-bottomed flask. The solution was refluxed for 2 h and then poured into a beaker containing some crushed ice. The white precipitate was washed with cold ethanol and distilled water, then dried in a vacumm (yield 80%). Recrystallization from acetone yielded single cystals suitable for X-ray analysis (m.p. 417.1–419.0 K). Analysis calculated: C 50.24, H 5.80, N 14.65, S 16.77, O 12.55%; found: C 49.3, H 5.60, N 14.13, S 15.86, O 11.983%.

 $D_{\rm r} = 1.346 {\rm Mg} {\rm m}^{-3}$

Cell parameters from 4016

Mo Ka radiation

reflections

 $\theta = 2.0-25.5^{\circ}$ $\mu = 0.31 \text{ mm}^{-1}$

T = 293 (2) K

 $R_{\rm int}=0.018$

 $\theta_{\rm max} = 25.5^{\circ}$

 $h = -10 \rightarrow 10$ $k = -5 \rightarrow 12$

 $l = -23 \rightarrow 24$

Block, colourless

 $0.42 \times 0.33 \times 0.25 \text{ mm}$

3505 independent reflections

 $w = 1/[\sigma^2(F_0^2) + (0.0429P)^2]$

-3

+ 0.9468*P*] where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} < 0.001$

 $\Delta \rho_{\rm max} = 0.33 \ {\rm e} \ {\rm \AA}^2$

 $\Delta \rho_{\rm min} = -0.27 \text{ e} \text{ Å}^{-3}$

3107 reflections with $I > 2\sigma(I)$

Crystal data

 $C_{15}H_{22}N_{3}O_{3}S^{+}\cdot NCS^{-}$ $M_{r} = 382.50$ Monoclinic, $P2_{1}/n$ a = 9.0195 (18) Å b = 10.402 (2) Å c = 20.289 (4) Å $\beta = 97.477$ (3)° V = 1887.3 (7) Å³ Z = 4

Data collection

Bruker SMART APEX CCD areadetector diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{min} = 0.882, T_{max} = 0.927$ 9351 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.125$ S = 1.173505 reflections 231 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Selected bond lengths (Å).

S1-C8	1.674 (2)	N2-C8	1.322 (3)
S2-C16	1.617 (3)	N2-C9	1.451 (3)
O1-C7	1.220 (3)	N3-C10	1.492 (3)
N1-C8	1.380 (3)	N3-C14	1.494 (3)
N1-C7	1.381 (3)	N3-C11	1.500 (3)

Table 2	
Hydrogen-bond geometry (Å	⊾, °).

$D - \mathbf{H} \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N2-H2A\cdots O1$	0.86	1.94	2.629 (3)	136
$C9-H9A\cdots S1$	0.97	2.63	3.123 (3)	112
$N3-H3A\cdots N4$	0.91 (2)	1.92 (2)	2.800 (4)	163 (2)
$N1 - H1A \cdot \cdot \cdot S1^{i}$	0.86	2.77	3.625 (2)	171
$C2-H2\cdots O2^{ii}$	0.93	2.60	3.357 (3)	139
C9−H9A···O3 ⁱⁱⁱ	0.97	2.40	3.206 (3)	141
$C10-H10A\cdots S2^{iv}$	0.97	2.82	3.786 (3)	172
$C15-H15C\cdots S2^{v}$	0.96	2.85	3.805 (4)	171

Symmetry codes: (i) -x + 1, -y + 1, -z; (ii) -x + 2, -y, -z; (iii) $-x + \frac{1}{2}$, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (iv) $-x + \frac{3}{2}$, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (v) x + 1, y, z.

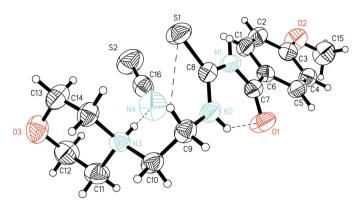


Figure 1

A view of the molecule of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

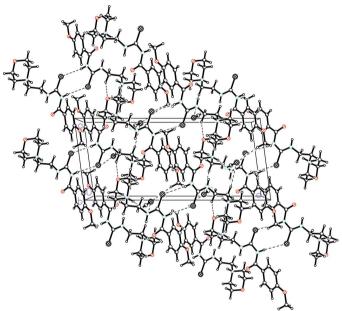


Figure 2

The crystal packing of the cation and anion of (I), viewed along the b axis. Dashed lines indicate hydrogen bonds.

After their location in a difference map, all H atoms except H3A were positioned geometrically at ideal positions and allowed to ride on the parent atom, with C-H = 0.93-0.96 Å and N-H = 0.86 Å, and with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm CH}_2 \text{ and NH})$ or $1.5U_{\rm eq}({\rm CH}_3)$. The protonated atom H3A, bonded to N3, was located in a difference map and refined freely.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

The authors thank the Malaysian Government for research grant IRPA No.09–02-02–0163.

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L. Orpen, A. G., & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659. Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany. Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.

Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Steinens (1990). SMART and SARTA. Steinens Analytical X-ray fits Inc., Madison, Wisconsin, USA.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7–13.
Yusof, M. S. & Yamin, B. M. (2005). Acta Cryst. E61, o2265–o2266.